Acta Crystallographica Section E

## Structure Reports

 OnlineISSN 1600-5368

S. Selvanayagam, ${ }^{\text {a }}$<br>D. Velmurugan, ${ }^{\text {a }}$<br>K. Ravikumar, ${ }^{\text {b }}$ S. Narasinga Rao, ${ }^{\text {c }}$ J. Jayashankaran, ${ }^{\text {d }}$ R. Rathna Durga ${ }^{\text {d }}$ and<br>R. Raghunathan ${ }^{\text {d }}$

${ }^{\text {a }}$ Department of Crystallography and Biophysics, University of Madras, Guindy Campus, Chennai 600 025, India, 'bLaboratory of X-ray Crystallography, Indian Institute of Chemical Technology, Hyderabad 500 007, India,
${ }^{c}$ University of Central Oklahoma, Edmond, Oklahoma, USA, and dDepartment of Organic Chemistry, University of Madras, Guindy Campus, Chennai 600 025, India

Correspondence e-mail: d_velu@yahoo.com

## Key indicators

Single-crystal X-ray study
$T=293 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.005 \AA$
$R$ factor $=0.084$
$w R$ factor $=0.194$
Data-to-parameter ratio $=15.6$
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

[^0]Printed in Great Britain - all rights reserved

## 8a,13b-cis-8,8,13-Trimethyl-11-phenyl-8,8a,9,13b-tetrahydropyrazolo[3",4"-b']thiapyrano $\left[5^{\prime}, 4^{\prime}: 3,4\right]$ pyrano[5,6-c]coumarin

In the title compound, $\mathrm{C}_{25} \mathrm{H}_{22} \mathrm{~N}_{2} \mathrm{O}_{3} \mathrm{~S}$, the dihydropyran ring adopts a half-chair conformation and the dihydrothiapyran ring adopts a sofa conformation.

Received 19 October 2004 Accepted 21 October 2004 Online 30 October 2004

## Comment

Coumarin derivatives show antimicrobial (Zaha \& Hazem, 2002) and vasorelaxant (Campos-Toimil et al., 2002) activities and serve as antiplatelet agents (Roma et al., 2003). These derivatives occurring in plants have different biological activities (Cisowski, 1983, 1984) and are used as dual inhibitors of acetylcholinesterase and monoamine oxidase (Bruhlmann et al., 2001). Recent results have shown that these derivatives act as potent and anti-HIV agents (Yu et al., 2003; Shikishima et al., 2001). In view of the above biological importance, the title compound, (I), was chosen for crystallographic study to determine its structure and conformation.

(I)

The title molecule (Fig. 1) consists of two benzene rings ( $A$ and $F$ ), one pyran ring $(B)$, one dihydropyran ring $(C)$, one dihydrothiapyran ring $(D)$ and one pyrazole ring $(E)$.

The geometry of the coumarin ring system is comparable to that observed in other coumarin derivatives (Chinnakali et al., 1998, 1999; Krishna et al., 2003). The bond distances agree well with the mean literature values (Allen et al., 1987). The sum of the angles at N 23 of the pyrazole ring $(E), 358.7^{\circ}$ is in accordance with $s p^{2}$-hybridization. In the coumarin moiety ( $A$ and $B$ ), the pyran ring $(B)$ is planar within 0.052 (3) $\AA$ and the dihedral angle between the weighted least-squares planes through the benzene and pyran ring is $2.0(1)^{\circ}$. The pyrazole ring $(E)$ is planar, with a maximum deviation of $-0.018(3) \AA$ for atom C21. The pyrazole ring $(E)$ and the phenyl ring $(F)$ subtend an angle of $51.2(1)^{\circ}$. The dihydropyran ring ( $C$ ) adopts a half-chair conformation, with the lowest asymmetry parameter of $\Delta C_{2}(\mathrm{C} 9-\mathrm{C} 8)=0.045(1)($ Nardelli, 1983). The dihydrothiapyran ring adopts a sofa conformation, with asymmetry parameters $\Delta C_{s}(\mathrm{C} 13)=0.021(1)$ and $\Delta C_{2}(\mathrm{C} 20-$ C13 $)=0.087$ (1) (Nardelli, 1983) .

## Experimental

To a solution of 3-methyl-5-(2-methylpropenylsulfanyl)-1-phenyl-1 H -pyrazole-4-carbaldehyde ( 1 mmol ) in ethanol was added an alcohol solution of 4-hydroxycoumarin (1 mmol) and ethylenediamine (1 drop)/diacetic acid (2 drops) (catalyst) at room temperature. The solution was refluxed for $3-4 \mathrm{~h}$. The completion of the reaction was evidenced by thin-layer chromatography. The solvent was removed in vacuo and subjected to column chromatography using petroleum ether and ethyl acetate (8:2) as eluant. Good quality crystals were obtained from a mixture of ethyl acetate and hexane (1:1) by slow evaporation.

## Crystal data

$$
\begin{aligned}
& \mathrm{C}_{25} \mathrm{H}_{22} \mathrm{~N}_{2} \mathrm{O}_{3} \mathrm{~S} \\
& M_{r}=430.51 \\
& \text { Monoclinic, } P 2_{1} / n \\
& a=10.1017(11) \AA \\
& b=10.2760(12) \AA \\
& c=20.187(2) \AA \\
& \beta=95.278(2)^{\circ} \\
& V=2086.7(4) \AA^{3} \\
& Z=4
\end{aligned}
$$

$$
\begin{aligned}
& D_{x}=1.370 \mathrm{Mg} \mathrm{~m}^{-3} \\
& \text { Mo } K \alpha \text { radiation }
\end{aligned}
$$

Cell parameters from 2365 reflections

$$
\theta=2.4-21.7^{\circ}
$$

$$
\mu=0.19 \mathrm{~mm}^{-1}
$$

$T=293$ (2) K
Block, colourless
$0.24 \times 0.20 \times 0.16 \mathrm{~mm}$

## Data collection

| Bruker SMART APEX | 3260 reflections with $I>2 \sigma(I)$ |
| :--- | :--- |
| $\quad$ diffractometer | $R_{\text {int }}=0.041$ |
| $\omega$ scans | $\theta_{\max }=28.0^{\circ}$ |
| Absorption correction: none | $h=-13 \rightarrow 13$ |
| 12360 measured reflections | $k=-13 \rightarrow 13$ |
| 4420 independent reflections | $l=-21 \rightarrow 25$ |

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.084$
$w R\left(F^{2}\right)=0.194$
$S=1.16$
4420 reflections
283 parameters
H -atom parameters constrained
$w=1 /\left[\sigma^{2}\left(F_{o}^{2}\right)+(0.0803 P)^{2}\right.$ $+0.9689 P]$
where $P=\left(F_{o}^{2}+2 F_{c}^{2}\right) / 3$
$(\Delta / \sigma)_{\max }<0.001$
$\Delta \rho_{\text {max }}=0.43 \mathrm{e}^{\circ}{ }^{-3}$
$\Delta \rho_{\min }=-0.24 \mathrm{e}^{-3}$
3260 reflections with $I>2 \sigma(I)$
$R_{\text {int }}=0.041$
$\theta_{\text {max }}=28.0^{\circ}$
$k=-13 \rightarrow 13$
$l=-21 \rightarrow 25$
$\rho_{\min }-0.24$ e A

Table 1
Selected geometric parameters ( $\left({ }^{\circ},{ }^{\circ}\right.$ ).

| S1-C19 | $1.737(3)$ | C14-O15 | $1.475(4)$ |
| :--- | ---: | :--- | :--- |
| S1-C20 | $1.801(4)$ | C19-N23 | $1.352(4)$ |
| C5-O6 | $1.368(4)$ | N22-N23 | $1.368(3)$ |
| O6-C7 | $1.392(4)$ | N23-C25 | $1.421(4)$ |
| C9-C10 | $1.465(4)$ |  |  |
| C13-C20-S1 | $117.0(2)$ | C19-N23-C25 | $127.3(3)$ |
| C19-N23-N22 | $110.8(2)$ | N22-N23-C25 | $120.6(3)$ |
|  |  |  |  |
| C19-C18-C21-C24 | $-173.0(3)$ | $\mathrm{C} 19-\mathrm{N} 23-\mathrm{C} 25-\mathrm{C} 26$ | $-57.4(4)$ |
| C24-C21-N22-N23 | $174.4(3)$ |  |  |

The H atoms were positioned geometrically and were treated as riding on their parent C atoms, with aromatic $\mathrm{C}-\mathrm{H}$ distances of $0.93 \AA$, methyl $\mathrm{C}-\mathrm{H}$ distances of $0.96 \AA$, ethylene $\mathrm{C}-\mathrm{H}$ distances of $0.97 \AA$ and methylene $\mathrm{C}-\mathrm{H}$ distances of $0.98 \AA$, and with $U_{\text {iso }}=$ $1.5 U_{\mathrm{eq}}(\mathrm{C})$ for methyl H and $1.2 U_{\mathrm{eq}}(\mathrm{C})$ for other H atoms.

Data collection: SMART (Bruker, 2001); cell refinement: SAINT (Bruker, 2001); data reduction: SAINT; program(s) used to solve


Figure 1
The molecular structure and atom-numbering scheme for (I). Displacement ellipsoids are drawn at the $30 \%$ probability level and H atoms are shown as small spheres of arbitrary radii.
structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEP-3 (Farrugia, 1997) and PLATON (Spek, 2003); software used to prepare material for publication: SHELXL97 and PARST (Nardelli, 1995).

SS thanks the Council of Scientific and Industrial Research (CSIR) for providing a Senior Research Fellowship. DV thanks the Department of Biotechnology (DBT) and University Grants Commission (UGC) for providing support under major research projects.

## References

Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. \& Taylor, R. (1987). J. Chem. Soc. Perkin Trans. 2, pp. S1-19.

Bruhlmann, C., Ooms, F., Carrupt, P. A., Testa, B., Catto, M., Leonetti, F., Altomare, C. \& Carotti, A. (2001). J. Med. Chem. 44, 3195-3198.
Bruker (2001). SAINT (Version 6.28a) and SMART (Version 5.625). Bruker AXS Inc., Madison, Wisconsin, USA.
Campos-Toimil, M., Orallo, F., Santana, L. \& Uriarte, E. (2002). Bioorg. Med. Chem. Lett. 12, 783-786.
Chinnakali, K., Fun, H.-K., Sriraghavan, K. \& Ramakrishnan, V. T. (1998). Acta Cryst. C54, 542-544.
Chinnakali, K., Fun, H.-K., Sriraghavan, K. \& Ramakrishnan, V. T. (1999). Acta Cryst. C55, 946-948.
Cisowski, W. (1983). Herba Pol. 29, 301-318.
Cisowski, W. (1984). Herba Pol. 30, 71-79.
Farrugia, L. J. (1997). J. Appl. Cryst. 30, 565.
Krishna, R., Selvanayagam, S., Yogavel, M., Velmurugan, D. \& Manikandan, S. (2003). Acta Cryst. E59, o667-o669.

Nardelli, M. (1983). Acta Cryst. C39, 1141-1142.
Nardelli, M. (1995). J. Appl. Cryst. 28, 659.
Roma, G., Braccio, M. D., Carrieri, A., Grossi, G., Leoncini, G., Grazia Signorello, M. \& Carotti, A. (2003). Bioorg. Med. Chem. 11, 123-138.
Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
Spek, A. L. (2003). J. Appl. Cryst. 36, 7-13.
Shikishima, Y., Takaishi, Y., Honda, G., Ito, M., Takfda, Y., Kodzhimatov, O. K., Ashurmetov, O. \& Lee, K. H. (2001). Chem. Pharm. Bull (Tokyo), 49, 877-880.
Yu, D., Suzuki, M., Xie, L., Morris-Natschke, S. L. \& Lee, K. H. (2003). Med. Res. Rev. 23, 322-345.
Zaha, A. A. \& Hazem, A. (2002). New Microbiol. 25, 213-222.


[^0]:    (C) 2004 International Union of Crystallography

